

Assessment/02



## Which of the following is a physical change?

- a) A piece of paper burns.
- b) An apple ripens.
- c) Gasoline is changed into carbon dioxide and water.
- d) Heat causes ice to change into water vapor.
- e) Sugar is digested.

Which of the following is NOT a chemical property?

- a) The density of a liquid
- b) The ability of iron to rust
- c) Flammability of gasoline
- d) Reactivity of acids with metals
- e) All of these are chemical properties.

**Which is the opposite of evaporation?**

- a) Sublimation**
- b) Deposition**
- c) Melting**
- d) Freezing**
- e) Condensation**

Which of the following is true for chemical changes?

- a) They are changes in state.
- b) They form new substances.
- c) They can usually be reversed.
- d) All of the above

Which one of the following choices includes only chemical changes?

- a) Decay, digestion, and burning
- b) Rusting, evaporation, and decay
- c) Melting, freezing, and digestion
- d) Corrosion, sublimation, burning
- e) Melting, evaporation, and sublimation

**Which of the following is a physical property?**

- a) A metal can react with acids.**
- b) Paper can be burned.**
- c) Salt crystals have a cube shape.**
- d) A metal can corrode by combining with oxygen.**
- e) Sugar can be digested**

**Physical properties are usually determined by observing how a substance reacts with another one.**

**a) True**

**b) False**



**Which of the following is a physical change?**

- a) Burning butane**
- b) Barbequing a steak**
- c) Decomposing water into hydrogen and oxygen gases**
- d) Breaking a piece of glass**
- e) Baking a cake**

**Which of the following statements is FALSE?**

- a) The freezing of water is a physical change.**
- b) The conversion of compounds into elements is a physical change.**
- c) Combining sodium metal and chlorine gas to form sodium chloride is a chemical change.**
- d) Photosynthesis is an example of a chemical change.**

**Boiling point is an example of a**

- a) physical change.**
- b) chemical change.**
- c) quantitative property.**
- d) physical property.**
- e) chemical property.**

